## **Surface Chemistry of Organic-Mineral Interactions**

Adsorption Mechanisms	Illustration	Some Relevant Minerals	Principal Organic Functional Groups Involved	A Few Significant Soil Properties
Hydrophobic Interactions (aka physical bonding, van der Waals interactions)	Polymer in solution  Exchangeable cation  B  Desorbed water molecules in the	any minerals with neutral microsites, eg smectite and kaolinite surfaces	uncharged, nonpolar groups eg aromatic, alkyl C	solution ionic strength amount SOM already attached to minerals
Hadaaaa Baadina	Adsorbed polymer Silicate layer			
Hydrogen Bonding	N $H-O-Si$ $O-Si$ $O-S$	any minerals with oxygen surfaces, eg kaolinite	amines, carbonyl, carboxyl, phenylhydroxyl, heterocycle N	water content
Protonation	Clay] <sup></sup> H <sub>3</sub> O <sup>+ −</sup> $\left\{ \begin{matrix} O \\ O \end{matrix} \right\}$ C − R	aluminosilicate edge sites Fe and Al oxides allophane and imogolite	amines, heterocyclic N, carbonyl, carboxylate	pH cations present water content
Ligand Exchange	Clay]-OH+ $\frac{-O}{HO}$ R $\longrightarrow$ Clay] $\frac{O}{O}$ R + H <sub>2</sub> O	aluminosilicate edge sites Fe and Al oxides allophane and imogolite	carboxylate, phenolate	structural metal cation water content

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Adsorption Mechanisms	Illustration	Some Relevant Minerals	Principal Organic Functional Groups Involved	A Few Significant Soil Properties
Ion Exchange permanent charge sites (cation exchange)	Colloid particle of soil + K Colloid + K + K	smectite, vermiculite, illite	amines, ring NH, heterocyclic N	cations present
pH-dependent charge sites (anion exchange usually, cation exchange rarely)	R-C 0 R-C 0	aluminosilicate edge sites Fe and Al oxides allophane and imogolite	carboxylate for anion exchange, amines, ring NH, heterocyclic N for cation exchange	pH solution ionic strength
Cation Bridging	R-0-0	smectite, vermiculite, illite	carboxylate, amines, carbonyl, alcoholic OH	cations present
Water Bridging	R—C-O-H H O-H······O·····M negative charge site	smectite, vermiculite, illite	amines, carboxylate, carbonyl, alcoholic OH	cations present water content

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